

# Heterogeneous reaction engineering



ECTS  
5 crédits



Hourly volume  
37h

## Introducing

Thermodynamics (I2BETH11)  
Fluid properties (I3BEPF12)

## Objectives

Know and explain the concepts of

- chemical catalysts
- heterogeneous reaction mechanisms and associated kinetic laws
- limiting step(s) in heterogeneous reactions
- apparent (overall) reaction rate
- dimensionless numbers (Hatta, Biot, Thiele, Weiss)
- effectiveness factor and enhancement factor

Establish an intrinsic kinetic law

Determine the limiting process(es) in a heterogeneous chemical reaction

Express dimensionless numbers used in heterogeneous reactions (Hatta, Biot, Thiele, Weisz) and explain their meaning

Express the apparent global rate of a chemical reaction depending on the working conditions

Select and design the most suitable reactor to perform a given reaction

Integrate and prioritize the mechanisms in order to model heterogeneous chemical reactors (batch or continuous)

## Practical info

### Location(s)

Toulouse

## Necessary prerequisites

Chemical reaction Engineering I (I2BERR12)  
Chemical reaction Engineering II (I3BERR12)  
Heat and mass transfer (I3BETF32)